STEREOISOMERS OF COLEONOL (FORSKOLIN) AND RELATED DITERPENOIDS

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Abstract: Regioselective Mitsunobu inversion of the hydroxyl groups of coleonol (forskolin) led to new, unnatural epimers as potential adenylate cyclase stimulant and pharmacodynamic agents.

Coleonol (Forskolin, $l\alpha$, 6β , 9α -trihydroxy- 7β -acetoxy-8, 13-epoxy-labd-14-en-11-one, 1), a labdane diterpenoid isolated¹ from the Indian medicinal plant <u>Coleus forskohlii</u> and a potential drug for glaucoma, congestive heart-failure and bronchial asthma, activates adenylate cyclase in unique manner by direct and reversible action on catalytic subunit of the enzyme in absence of guanine nucleotide stimulating protein². Forskolin, owing to its highly oxygenated structure and novel pharmacodynamic action, has emerged as a highly attractive target for synthetic investigations².

During our studies on the physiological effects of Coleonol (Forskolin) and its analogues, we decided to prepare new and unnatural stereoisomers (epimers) of 1 to correlate biological activity with the stereochemical orientation of various hydroxyl groups. Herein we report the synthesis of various 1-epi- and 7-epi coleonols (forskolins) and 6,7-epoxy-6,7-dideoxycoleonol using versatile Mitsunobu reaction³ which has become a useful tool in organic synthesis with stereochemical outcome being almost invariably a clean inversion of configuration of secondary hydroxyl centres. The reaction of 1 (1 mmol) with triphenylphosphine (TPP, 3 mmol), diethylazodiacarboxylate (DEAD, 3 mmol) and benzoic acid (1.1 mmol) in dry THF at room temperature for 36 h gave compound 2 after chromatography and characterized as 18-benzoyloxy-1-deoxycoleonol (yield 45%). The PMR spectra of 2 showed $\frac{4}{dd}$ at δ 5.75 (J_{ax/ax} = 10 Hz, J_{ax/eq} = 2Hz) assigned to 1- α -axial proton coupled with two neighbouring C-2ax and C-2eq protons. Debenzoylation of 2 with methanolic NaOMe at r.t. yielded 1-epi-7-deacetylcoleonol (3) which on selective acetylation (Ac20, Py, 0-5°) gave i-epi coleonol (4). The stereochemistry of C,-OH of 4 was assigned as β -equatorial by PMR⁵ which showed <u>dd</u> at 3.95 (J_{ax/ax} = 9.5 & J_{ax/eq} = 2.0 Hz). A recently modified Mitsunobu procedure⁶ to obtain equatorial vs axial-hydroxyl selectivity using trifluoroacetic acid (TFA) as nucleophile and sodium benzoate as catalyst was used to synthesize 7α -axial epimer of 1. The 7-deacetylcoleonol (5) prepared by deacetylation of 1, on Mitsunobu reaction using TFA (1.1 mole equivalent) and NaOBz afforded 7-deacetyl-7-epi-trifluoro-

acetylcoleonol (6) in 60% yield. The PMR of 6 showed⁷ <u>d</u> at 5.50 (J = 2 Hz) for 7- β -equatorial C<u>H</u> proton. The hydrolysis of 6 by refluxing in MeOH led to 7-deacetyl-7-<u>epi</u>-coleonol (7). The reaction of coleonol-B (6-acetyl-7-deacetylcoleonol, 8) with TPP-DEAD complex using TFA and NaOBz gave 6-acetyl-7-deacetyl-7-<u>epi</u>-



trifluoroacetylcoleonol (9) which on selective hydrolysis (reflux in MeOH) gave

7-epi coleonol-B (10). Further hydrolysis (NaOMe in MeOH at r.t.) 9 gave 7. An interesting product, 6,7-epoxy-6,7-dideoxycoleonol (11) was formed when 7-deacetylcoleonol (5) was reacted with TPP-DEAD complex in dry THF in absence of acidic component. The result indicated



the inversion at C-7 being successful but followed by an intramolecular nucleophilic displacement by free hydroxyl at C-6. The stereochemistry of epoxide ring was ascertained^o to be 68-78-oriented as determined by PMR spectrum of 11.

Conclusively, it is possible to regioselectively invert the configuration of 1- and 7-OH of forskolin using Mitsunobu reaction to prepare new and unnatural epimeric compounds as potential pharmacodynamic agents. The 6- and 9-hydroxyl groups being much more hindered did not react. The biological activity of the epimeric forskolins will be published separately.

References and Notes

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- PMR data of 2. 7.50 (m, 5H, COPh), 6.00 (dd, 1H, H-14, J = 10 & 17), 5.75 (dd, 1H, H-1ax, 4. J = 10 & 2, 5.50 (d, 1H, H-7, $\overline{J} = 4$), 5.24 (dd, 1H, H-15, J = 17 & 1.5), 4.48 (m, 1H, H-6), 4.93 (dd, 1H, \overline{H} -15', J = 10 & 1.5), 3.18 (d, 1H, H-12, J = 17), 2.50 (d, 1H, H-12,
- H-6), 4.93 (dd, 1H, H-15', J = 10 & 1.5), 3.18 (d, 1H, H-12, J = 17), 2.50 (d, 1H, H-12', J = 17), 2.15 (s, 3H, COCH₂), 1.00, 1.10, 1.27, 1.30, 1.60 (all s, 5 x CH₂). PMR data of 4. 6.13 (dd, 1H, H-14, J = 10 & 17)), 5.21 (dd, 1H, H-15, J = 17 & 1.5), 4.99 (dd, 1H, H-15, J = 10 & 1.5), 3.95 (dd, 1H, H-1ax, J = 9.5 & 2), 4.40 (m, 1H, H-6), 5.50 (d, 1H, H-7, J = 4), 3.18 (d, 1H, H-12, J = 17), 2.50 (d, 1H, H-12, J = 17), 2.10 (d, 1H, H-5), J = 3), 2.15 (s, 3H, COCH₂), 1.0, 1.05, 1.25, 1.40, 1.60 (all s, 5 x CH₂). M. Varasi, K.A.M. Walker and M.L. Maddox, <u>J. Org. Chem.</u>, 52, 4235 (1987). PMR data of 6. 6.00 (dd, 1H, H-14, J = 10 & 17), 5.21 (dd, 1H, H-15, J = 17 & 1.5), 4.99 (dd, 1H, H-15, J = 10 & 1.5), 5.50 (d, 1H, H-7, J = 2), 4.55 (m, 1H, H-1), 4.48 (m, 1H, H-6), 3.15 (d, 1H, H-12, J = 17), 2.50 (d, 1H, H-12, J = 17), 2.30 (d, 1H, H-5), 0.95, 1.05, 1.25, 1.30. 1.55 (all s, 5 x CH₂). 5.
- 6.
- 1.25, 1.30, 1.55 (all s, 5 x CH). PMR data of 11. 6.00 (dd, 1H, H-14, J = 10 & 17), 5.15 (dd, 1H, H-15, J = 17 & 1.5),
- 8. 5.00 (dd, 1H, H-15, $J = \overline{10}$), 4.20 (m, 1H, H-1), 3.30 (d, 1H, H-12, J = 15), 2.40 (d, 1H, H-12', J = 15), 2.55 (d, 1H, H-7, J = 3.5), 2.15 (dd, 1H, H-6, J = 3.5 & 1.9), 2.05 (d, 1H, H-5, J = 1.9, 1.20, 1.21, 1.40, 1.50, 1.60 (all s, 5 x CH₃).

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